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### (54) IMPROVEMENTS IN AND RELATING TO MICROELECTRODES AND AMPEROMETRIC ASSAYS.

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**Description****Technical Field**

This invention relates to an improved method of making an electrode (in particular a microelectrode), an electrode made by the method, an assay method for the determination of the presence of a redox species (e.g. a heavy metal) in liquid carrier media using the electrode and redox species detection equipment designed to utilise the assay method.

**Background Art**

Microelectrodes provide many advantages in electrochemistry and electroanalysis (some of which are discussed in Wightman, R.M., *Anal. Chem.*, 1982, 54, 2532) in particular the possibility of electrolysis in resistive media or dilute solutions, and of microanalysis with improved sensitivity and precision. For these reasons a great deal of effort has been dedicated to establish a way of fabricating a regular microscopic disc array electrode. Any proposed method of fabrication must ensure that all the distinctive features of individual microelectrodes are retained, and in addition circumvent the problems associated with the measurement of extremely small currents (pA-nA), observed for single microdisc electrodes.

Many approaches have been proposed for the construction of a microdisc array electrode including photolithographic methods (see e.g. Osteryoung, J. and Hempel, T., *J. Electrochem. Soc.*, 1986, 133, 757-760 and W. Siu and R.S.C. Cobbold, *Med. & Biol. Eng.*, 1976, 14, 109), as well as the use of micropore membrane matrices (see e.g. Wang, J., *J. Electrochem. Soc.*, 1988, 249, 339-345, and Cheng, F., *Anal. Chem.*, 1989, 61, 762-766). The former approach failed because of adhesion problems between the insulating photoresist layer and the metal substrate. The latter method of fabrication was self defeating because, although the size of the pore is well characterised, the pore density and distribution were undefined and the electrode ultimately irreproducible.

**Summary of the Invention**

According to the first aspect of this invention a method of making a microelectrode comprising a layer of electrically insulating material having an array of apertures formed therein and electrically conducting material visible through the apertures, is characterised in that the apertures are formed by photo-ablation.

Suitably the layer is a sheet of flexible plastics material (suitably a polyester or other polymer film) and the apertures are formed by photo-ablation. A particularly preferred form of electrode made by the method of this invention is a microdisc array electrode

in which the apertures are of the same size and shape and are distributed over the sheet in a regular array. In the field of microdisc electrodes the sheet can be of a thickness in the range 2 µm to 500 µm and the apertures are suitably of diameter (or transverse dimension if not circular) of between 5 µm and 100 µm. A hexagonal array of 10 µm diameter apertures with edge/edge separations of 100 µm between the apertures in the array has been found to be one convenient arrangement offering many useful commercial applications.

An alternative method of making the electrode involves supporting a layer of electrode material on a substrate, laying a thin polymer film onto the electrode material (e.g. by vapour polymerisation) and then drilling apertures by photo-ablation through the polymer film. The layer of electrode material is conveniently thick-film printed (to any desired pattern) on the substrate. This alternative method has a particular advantage when the electrode material includes carbon, since the photo-ablation used to form the apertures can vitrify the carbon in the areas of electrode material exposed by the photo-ablation.

As well as microdisc arrays the invention extends to microband arrays, a microband array electrode comprising a set of spaced-apart slot-shaped apertures exposing narrow strips (e.g. parallel strips) of a common electrode material.

The formation of the array of apertures by a photo-ablation technique has proved to be particularly convenient and typically involves photographically creating spaced-apart ablative areas on the sheet interspersed by more resistant non-ablative regions, and then removing the sheet material only over the ablative areas using a high power source of radiation (e.g. from an excimer laser). Using photo-ablation, means that an exceedingly accurate array of apertures of any required size can be produced using well established techniques.

The conducting material is suitably a thixotropic paste based on carbon or metallic particles (e.g. platinum or gold) applied sufficiently thickly so that where the paste is applied to a pre-drilled sheet of plastics material each aperture formed in the non-conducting sheet is substantially filled with conducting particles.

A pre-drilled coated sheet can be bonded, conducting side down, on a substrate of sufficient thickness to give the required rigidity to the sandwich thus produced. The alternative thick film method which drills apertures through to the electrode material already has a suitable support substrate. The substrate can be of electrically conducting or non-conducting material, and can incorporate suitable electrical conductors for applying potentials to and/or leading currents to/from the one or more conducting areas exposed in the apertures.

From the above, it will be appreciated that the

fabrication method proposed in this first aspect of the invention is essentially a three dimensional thick film process. The procedure circumvents all the past difficulties including the aforementioned, and is based on principles of photo-machining of polymer films by laser photo-ablation to produce matrices of well defined aperture size, distribution, density and geometry. The process can fabricate array patterns in support matrices of thickness 2 µm to 500 µm, and the apertures can be filled with the required conducting particles. The filled sheet can then be mounted by encapsulation on an inert substrate e.g. by a neat sealing process.

A microelectrode made by the method of the invention represents a further aspect of this invention. Initial trials have suggested that such electrodes have useful applications in electroanalysis, high resolution electrochemistry, pathogen testing techniques, antibody or enzyme reactions, and bioelectrochemical assay methods.

The following list details the important characteristics of an electrode made by the method of the invention which underline its use in electroanalysis and application to chemical sensors. It should be appreciated that the method of fabrication here discussed is uniquely controllable in terms of device specification and this has not been found to be true for earlier attempts at fabrication.

1. Uniformity of electrode shapes within an array, size may range from 5 µm to 50 µm in diameter. A reproducible manufacturing method produces identical array electrodes and is essential for reliable analytical application. The small aperture size produces an amperometric signal with improved analytical properties such as detectability, sensitivity and precision even in the presence of dissolved oxygen in the analyte.
2. Variable array size, from a few apertures (say six) to many thousands. The number of apertures within the array will depend on application: current amplification increases linearly with the number of isolated electrodes.
3. Arrays may be patterned: e.g. hexagonal, square etc.
4. The spacing between conducting areas can be varied at will. The ability to set the aperture separation is a very advantageous feature. For example, the separation can be made very small (a few µm) for the monitoring of catalytic current.
5. Arrays of electrodes can be made from different conducting materials (e.g. platinum, gold, carbon etc).
6. The composition of electrode material can be varied (e.g. ligands and complexing agents can be incorporated in the conducting material to add chemical selectivity to an electroanalysis to be performed with the electrode array).
7. Apertures or groups of apertures within an ar-

ray can be individually addressed to a set electric potential and therefore the electrode array could be used in multicomponent determination simultaneously.

5 8. The electrode design is ideally suited to chemical immobilisation of reactants (enzyme or antibody) onto the insulating area by covalent bonding hence allowing direct application to biosensor and biochemical assay technology.

10 A particularly promising application involves the use of an electrode made by the method of this invention for redox species analysis using Voltammetry.

15 Accurate determination of redox species such as pesticides or heavy metals in water is important to the monitoring and control of environmental pollution. The type of information required for comprehensive environment studies not only includes a single sampled measurement of a specific redox species (e.g. metal) at a given place and time, but should include extensive distribution and fluctuation data on major pollutants in order to ascertain origin.

20 The techniques presently employed by consultant analysts for the determination of heavy metals in water samples include the laboratory based techniques. Atomic Spectroscopy (Absorption and Emission) and Polarography (Pulse and Derivative). These conventional analytical methods are by nature timely and consequently expensive, relying on sampling procedures of a given water supply.

25 30 This invention also relates to an analytical method and device which need not be laboratory bound. In one embodiment, the device is a multi-heavy metal ion detector directed essentially at in-the-field measurement which may be used as a portable hand-held device or form part of a remote sensing network.

An analytical method based on this aspect of the invention is electrochemically based and relies on the properties of a microelectrode according to this invention.

35 40 According to this aspect of the invention an assay method for a target redox species analysis in a liquid carrier comprises moistening an absorbent medium containing a dry reactant for the species with the liquid carrier, the absorbent medium being sandwiched between two electrodes, one of which is a microelectrode according to this invention, the moistening process being monitored by measuring the conductance between the two electrodes and the concentration of the target redox species being determined by measuring the current flowing between the two electrodes when applying a varying potential therebetween.

45 50 Suitably a programmed voltage scan (ramp or step formed) is used for the concentration measurement.

55 The target redox species could be, inter alia, a drug, a bioproduct or a heavy metal ion.

According to a further aspect of the analytical method aspect of the invention an assay method for

a trace redox species analysis in a liquid carrier comprises moistening an absorbent medium containing a dry reactant for the species with the liquid carrier, the absorbent medium being sandwiched between two electrodes, one of which is a microelectrode according to this invention, applying a first potential across the electrodes to cause any species drawn into the medium and reacted with the reactant therein to migrate ions onto the conductive areas of the microelectrode, applying a second potential of opposite polarity across the electrodes to strip the said ions from the microelectrode and determining the nature of the "stripped" ion from a knowledge of the redox potential required to effect stripping.

A particularly useful application of the assay method of this invention for trace redox species is in the determination of which heavy metal ions are present in a sample of water, the dry reactant in the absorbent medium being a mercury salt. The method of analysis in this case is an adaption of anodic stripping voltammetry.

In its equipment aspect the invention comprises an assay unit which includes a pad of dry absorbent material incorporating the required dry form redox reagent sandwiched between a microelectrode according to this invention and a counter electrode, the sandwich being mounted on a support means that includes a contact to at least one of the electrodes.

Suitably the support means is a non-conducting (e.g. glass) plate having two contacts, one leading to each of the different electrodes. In the case of an assay device for heavy metal detection, the redox reagent can be a mercury salt and where testing on waters with very low electrical conductivity is in mind, a readily ionisable salt (such as sodium chloride).

An assay device in accordance with this aspect of the invention can be used as a disposable component which is connected to (e.g. plugged into) a suitable potentiostat. A fully portable testing equipment is available in this manner, with testing simply requiring a wetting of the absorbent pad of a fresh device and subsequent application of the appropriate potentials to the electrodes using the potentiostat to carry out the required Voltammetry or Anodic Stripping Voltammetry.

#### Brief Description of Drawings

The various aspects of this invention will now be further described, by way of example, with reference to the accompanying drawings, in which:

Figure 1 is a much enlarged sectional view of part of a microdisc array electrode according to the invention,

Figure 2 is a plan view from above of the part shown in Figure 1,

Figure 3 is a much enlarged sectional side view of part of a microband array electrode according

to the invention,

Figure 4 is a plan view from above of the part shown in Figure 3,

Figure 5 is a schematic view of laser equipment for forming an array of apertures in a component of a micro-electrode,

Figure 6 is a side elevational view of an assay device in accordance with a further aspect of the invention,

Figure 7 is a schematic sectional view through part of the device of Figure 6 with the vertical scale exaggerated for ease of viewing, and

Figure 8 is a graph showing the applied potential for a trace element detected with a device such as that shown in Figures 6 and 7 by stripping analysis.

#### Description of Preferred Embodiments

The microdisc array electrode partially shown in Figures 1 and 2 comprises a substrate 1 on which a perforated sheet 2 of non-conducting material coated on one side with a layer 3 of conducting material is applied.

The perforations are regularly spaced equally sized apertures 4 drilled by photo-ablation and forming an array over the entire effective area of the microdisc array electrode.

In a typical case the substrate 1 would be a sheet of polyvinylchloride of a thickness of 500 µm, the sheet 2 a polyester sheet of a thickness of 100 µm and the layer 3 a cured carbon paste of a cured thickness of 500 µm. The array shown in Figure 2 has circular discs of 10 micron diameter in a 100 µm spaced hexagon pattern. Other geometries such as those leading to microband electrodes are also envisaged. Since each aperture is filled with the conducting paste, each filled aperture forms a conducting area surrounded by a non-conducting region. All exposed conducting areas (or discs) can be connected to the same layer 3 but it will be appreciated that where end uses require different potentials to be applied to different discs or different groups of discs, the layer 3 can be appropriately divided and connected to its own supply lead. The supply lead has not been shown in Figures 1 and 2 but such a lead can be seen at 11 in Figure 6.

Figures 3 and 4 correspond to Figures 1 and 2 but show a microband array electrode the apertures 4 now being narrow slits. The micro-array electrode shown in Figure 3 was made by thick-film printing conducting electrode material 3 onto an electrically insulating substrate 1 in the required pattern. A layer 2 of polymer film is then vapour polymerised over the printed material 3 and subsequently drilled using laser photo-ablation to produce the apertures 4. With carbon in the material 3 vitrified carbon is formed in the apertures 4 giving the electrode high durability

and good electrical properties.

Photo-ablation of the required aperture pattern is schematically illustrated in Figure 5 where 30 indicates a high power laser source (e.g. a Lambda Physik LPX 2051 unit), 31 a reducing lens and 2 the plastics film requiring to be drilled. The film 2 can have an apertured mask printed thereon whereby only the film exposed through the apertures in the mask is available for ablation or a mask 32 (e.g. of metal) can be located in the optical path anywhere between the laser source 31 and the film 2. A metal mask contacting the film 2 can also be used.

The minimum size of the apertures 4 that can be produced in a plastics film are expected to be 1 to 2  $\mu\text{m}$  but they can be as large as desired.

The film 2 could be poly-para-xyllylene (Parylene N) or other CH materials (such as polystyrene or polyethylene for example).

As examples of suitable laser sources may be mentioned a KrF laser (wavelength 248 nm) of fluence in the range 0.2 to 0.4 J/cm<sup>2</sup> and (depending on the polymer film 2 being used) ArF (wavelength 193 nm) and XeCl (wavelength 308 nm) lasers of fluences in the range 0.1 to 1.0 J/cm<sup>2</sup>.

For contact imaging using a metal mask 32 in contact with a Parylene N film of 2  $\mu\text{m}$  thickness, a KrF laser of 300 mJ output (a Questek 2440 unit) can be used generating a fluence at the mask of 0.3 J/cm<sup>2</sup> over an area of 1 cm<sup>2</sup>. Using 25 shots of the source 30, an array of 15  $\mu\text{m}$  diameter apertures 4 on 100  $\mu\text{m}$  hole centres was accurately formed in the film 2.

For projection imaging, a metal mask 32 with an array of 45  $\mu\text{m}$  diameter holes on 300  $\mu\text{m}$  hole centres was located in the beam from a 600 mJ KrF laser and projected down onto a 2  $\mu\text{m}$  thick Parylene N film with a three times reduction of image in the lens 31. The fluence at the mask was 0.044 J/cm<sup>2</sup> and at the film it was 0.4 J/cm<sup>2</sup>. Again 25 shots of the source 30 were required to drill the accurate array of apertures 4 in the film 2.

The device shown in Figures 6 and 7 comprises a rectangular substrate 10 of non-conducting material on which a microelectrode 12 has been provided. Adhered over the electrode 12 is a pad 13 of absorbent material and overlying the pad 13 is a counter electrode 14. The electrode 14 can be applied by printing and will have an area larger than the sum of the apertures of the electrode 13. Conducting leads 11 and 15 connect contact areas 16 and 17 to the two electrodes 12 and 14.

Conveniently, the areas 16 and 17 are positioned so that slipping the substrate 10 into a holder (shown dotted at 20) which is connected to an electronic unit 21 by a cable 22, automatically places the device in circuit with the electronic equipment in the unit 21. The unit 21 can comprise a miniaturised microprocessor with control potentiostat incorporating a sensitive microammeter.

For testing for the presence of heavy metals (e.g. Pb, Cd, Fe, Zn) in water, the pad 13 will include some particles of common salt and mercurous chloride (say 10% w/w of NaCl and 5% w/w of Hg<sub>2</sub>Cl<sub>2</sub>) or other dried reactants necessary to provide conductivity and oxidised mercury for an electroplating operation. The micro-conducting areas of the electrode 12 will be carbon and the electrode 14 will be of silver/silver chloride. Dipping the pad 13 into a sample of the water to be tested, will draw water into the pad 13 by capillarity action where it will dissolve the salt and redox reagent forming a conductive solution between the electrodes 12 and 14.

For voltammetric analysis, current is passed with electrode 12 as working electrode and electrode 14 as a controlled voltage secondary electrode. The electrochemical reaction on electrode 12, i.e. oxidation or reduction of the target redox species is followed by classical microelectrode electroanalytical methodology.

For stripping analysis, current is passed with electrode 14 as anode to plate mercury and any heavy metal present onto the micro-conductive areas of the electrode 12. Plating current is maintained by hemispherical mass transport inherent to the micro-array geometry for long enough (typically 2 or 3 minutes) to plate a detectable amount of the heavy metal onto the conducting areas.

By virtue of the fact that the microelectrodes are cheap enough to be disposed of after a single use, "memory" problems, such as can occur with laboratory-based equipment, will be avoided since after use for one analysis, the electrode will be discarded.

The polarity of the applied potential is now reversed so that electrode 12 serves as anode and the polarity is increased as a plot is taken of current against voltage. One example of such a plot is shown in Figure 8 which shows the stripping of lead cations from mercury coated micro-conductive areas.

It will be appreciated, therefore, that the invention makes possible an assay device which comes in two sections; a sensing unit consisting of a disposable or reusable film cartridge driven by a pocket size computerised potentiostat which comprises the control and signal handling unit. The output of the instrument is a current which is proportional to the concentration of the trace metal ion at a potential specific to the ion itself.

The electroanalysis method described hereabove can be easily adapted to enzyme chemistry or/and immunochemistry for use as a biosensor. Indeed, enzymes or/and antibodies can be covalently immobilized on the perforated polymer sheet 2 in Figure 1 and electron carrier species often called mediators can be detected on the microelectrode array in a very efficient manner due to the close proximity between the electrode and the immobilised species leading to feedback diffusion enhancement.

It should be stressed that the sensitivity and accuracy of the method and apparatus described herein, are at least comparable to the equivalent laboratory-based methods and equipment.

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### Claims

1. A method of making a microelectrode comprising a layer of electrically insulating material having an array of apertures formed therein and electrically conducting material visible through the apertures, characterised in that the apertures are formed by photo-ablation.
2. A method according to claim 1, characterised in that the insulating layer is a plastics film and the conducting material is applied to one side of the film after the apertures have been formed therein.
3. A method according to claim 1, characterised in that the insulating layer is applied over conducting material supported on an electrically insulating substrate and the apertures are then formed in the insulating layer to expose the conducting material.
4. A microelectrode made by the method of claim 1, characterised in that the layer of insulating material has a thickness in the range of 2 to 500 µm and each aperture has a maximum transverse dimension of between 5 and 100 µm.
5. A method for a target redox species analysis in a liquid carrier comprising moistening an absorbent medium containing a dry reactant for the species with the liquid carrier, the absorbent medium being sandwiched between two electrodes, characterised in that one of the electrodes is a microelectrode made by the method according to claim 1, the moistening process being monitored by measuring the conductance between the two electrodes and the concentration of the target redox species being determined by measuring the current flowing between the two electrodes when applying a varying potential therebetween.
6. A method according to claim 5, characterised in that a programmed voltage scan is used for the concentration measurement.
7. An assay method for a trace redox species analysis in a liquid carrier comprising moistening an absorbent medium containing a dry reactant for the species with the liquid carrier, the absorbent medium being sandwiched between two electrodes, characterised in that one of the electrodes is

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a microelectrode made by the method according to claim 1, and in that a first potential is applied across the electrodes to cause any species drawn into the medium and reacted with the reactant therein to migrate ions onto the conductive areas of the microelectrode, applying a second potential of opposite polarity across the electrodes to strip the said ions from the microelectrode and determining the nature of the stripped ion from a knowledge of the redox potential required to effect stripping.

8. An assay unit, characterised in that it includes a pad of dry absorbent material sandwiched between a microelectrode as claimed in claim 1 and a counter electrode, the absorbent material incorporating a dry-form redox reagent and the sandwich being mounted on a support means that includes a contact to at least one of the electrodes.
9. An assay unit as claimed in claim 8, characterised in that the support means is a non-conducting plate having two contacts, one leading to each of the different electrodes.
10. An assay unit as claimed in claim 9, characterised in that the redox reagent is a mercury salt.

### Patentansprüche

1. Verfahren zur Herstellung einer Mikroelektrode mit einer Schicht aus elektrisch isolierendem Material, in der eine regelmäßige Anordnung von Öffnungen ausgebildet ist und elektrisch leitendes Material durch die Öffnungen sichtbar ist, dadurch gekennzeichnet, daß die Öffnungen mittels Abtragung durch Lichtimpulse gebildet werden.
2. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß es sich bei der Isolationsschicht um eine Kunststoffsicht handelt und das leitfähige Material auf eine Seite der Schicht nach Ausbildung der Öffnungen in derselben aufgetragen wird.
3. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß die Isolationsschicht auf von einem elektrisch isolierenden Träger getragenes leitfähiges Material aufgetragen wird und die Öffnungen dann in der Isolationsschicht ausgebildet werden, um das leitfähige Material freizulegen.
4. Mikroelektrode, hergestellt nach dem Verfahren des Anspruch 1, dadurch gekennzeichnet, daß die Schicht aus isolierendem Material eine Dicke im Bereich von 2 bis 500 µm aufweist und jede

Öffnung eine maximale Querabmessung zwischen 5 und 100 µm aufweist.

5. Verfahren zur Analyse auf eine zu bestimmende Redoxspezies in einem flüssigen Träger, bei dem ein einen trockenen Reaktionspartner für die Spezies enthaltendes saugfähiges Medium mit dem flüssigen Träger befeuchtet wird, wobei das saugfähige Medium zwischen zwei Elektroden eingelagert ist, dadurch gekennzeichnet, daß es sich bei einer der Elektroden um eine nach dem Verfahren gemäß Anspruch 1 hergestellte Mikroelektrode handelt, wobei der Befeuchtungsprozeß durch Messen der Leitfähigkeit zwischen den zwei Elektroden überwacht und die Konzentration der zu bestimmenden Redoxspezies durch Messen des Stroms bestimmt wird, der bei Anlegen eines wechselnden Potentials zwischen den beiden Elektroden zwischen diesen fließt.
10. Verfahren nach Anspruch 5, dadurch gekennzeichnet, daß zur Messung der Konzentration eine programmierte Spannungsabtastung erfolgt.
15. Bestimmungsverfahren zur Analyse auf eine in Spuren vorliegende Redoxspezies in einem flüssigen Träger, bei dem ein einen trockenen Reaktionspartner für die Spezies enthaltendes saugfähiges Medium mit dem flüssigen Träger befeuchtet wird, wobei das saugfähige Medium zwischen zwei Elektroden eingelagert ist, dadurch gekennzeichnet, daß es sich bei einer der Elektroden um eine nach dem Verfahren gemäß Anspruch 1 hergestellte Mikroelektrode handelt, und daß ein erstes Potential über die Elektroden angelegt wird, um zu bewirken, daß etwa von dem Medium aufgenommene und mit dem Reaktionspartner darin umgesetzte Spezies als Ionen auf die leitfähigen Gebiete der Mikroelektrode wandern, wobei ein zweites Potential entgegengesetzter Polarität über die Elektroden angelegt wird, um die Ionen von der Mikroelektrode abzulösen, und die Art der abgelösten Ionen aus der Kenntnis des zur Ablösung erforderlichen Redoxpotentials herbestimmt wird.
20. Bestimmungsgerät, dadurch gekennzeichnet, daß es ein zwischen einer Mikroelektrode nach Anspruch 1 und einer Gegenelektrode eingelagertes Kissen aus trockenem saugfähigem Material beinhaltet, wobei in das saugfähige Material ein Redoxreagens in trockener Form eingeschüttet ist und die Schichtanordnung auf einer Trägervorrichtung befestigt ist, die einen Kontakt mit mindestens einer der Elektroden beinhaltet.
25. Bestimmungsgerät nach Anspruch 8, dadurch gekennzeichnet, daß es sich bei der Trägervor-

richtung um eine nichtleitende Platte mit zwei Kontakten handelt, von denen einer zu jeder der verschiedenen Elektroden führt.

5. Bestimmungsgerät nach Anspruch 9, dadurch gekennzeichnet, daß es sich bei dem Redoxreagens um ein Quecksilbersalz handelt.
10. Revendications
15. 1. Méthode de fabrication d'une microélectrode comprenant une couche de matériau isolant électriquement, ayant un réseau d'ouvertures formées dans celui-ci et un matériau conducteur électriquement, visible à travers les ouvertures, caractérisée en ce que les ouvertures sont formées par photo-ablation.
20. 2. Méthode selon la revendication 1, caractérisée en ce que la couche isolante est un film de plastique et en ce que le matériau conducteur est appliqué sur un côté du film après la formation des ouvertures en son sein.
25. 3. Méthode selon la revendication 1, caractérisée en ce que la couche isolante est appliquée sur du matériau conducteur supporté sur un substrat isolant électriquement et en ce que les ouvertures sont alors formées dans la couche isolante pour exposer le matériau conducteur.
30. 4. Microélectrode fabriquée par l'intermédiaire de la méthode de la revendication 1, caractérisée en ce que la couche de matériau isolant a une épaisseur dans le domaine de 2 à 500 µm et en ce que chaque ouverture a une dimension transversale maximum comprise entre 5 et 100 µm.
35. 5. Méthode en vue d'une analyse d'une espèce redox cible dans un vecteur liquide, comprenant l'humidification d'un milieu absorbant contenant un réactif sec pour l'espèce avec le vecteur liquide, le milieu absorbant étant pris en sandwich entre deux électrodes, caractérisée en ce que l'une des électrodes est une microélectrode fabriquée par l'entremise de la méthode selon la revendication 1, le processus d'humidification étant surveillé par la mesure de la conductance entre les deux électrodes et la concentration de l'espèce redox cible étant déterminée par la mesure du courant circulant entre les deux électrodes quand on applique un potentiel variable entre elles.
40. 6. Méthode selon la revendication 5, caractérisée en ce qu'un balayage de voltage programmé est utilisé pour la mesure de la concentration.
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7. Méthode d'essai pour une analyse d'une espèce redox sous forme de traces dans un vecteur liquide, comprenant l'humidification d'un milieu absorbant contenant un réactif sec pour l'espèce avec le vecteur liquide, le milieu absorbant étant pris en sandwich entre deux électrodes, caractérisée en ce que l'une des électrodes est une microélectrode fabriquée par l'entremise de la méthode selon la revendication 1, et en ce qu'un premier potentiel est appliqué aux électrodes pour contraindre toute espèce, attirée dans le milieu et ayant réagi avec le réactif qui s'y trouve, à faire migrer des ions dans les zones conductrices de la microélectrode, l'application d'un deuxième potentiel de polarité opposée aux électrodes pour détacher lesdits ions de la microélectrode et la détermination de la nature de l'ion détaché à partir de la connaissance du potentiel redox nécessaire pour effectuer le détachement.
8. Unité d'essai, caractérisée en ce qu'elle comprend un tampon de matériau absorbant sec pris en sandwich entre une microélectrode comme revendiquée dans la revendication 1 et une contre-électrode, le matériau absorbant incorporant un réactif redox sous forme sèche et le sandwich étant monté sur un moyen de support qui comprend un contact à au moins l'une des électrodes.
9. Unité d'essai comme revendiquée dans la revendication 8, caractérisée en ce que le moyen de support est une plaque non conductrice ayant deux contacts, l'un conduisant à chacune des différentes électrodes.
10. Unité d'essai comme revendiquée dans la revendication 9, caractérisée en ce que le réactif redox est un sel de mercure.

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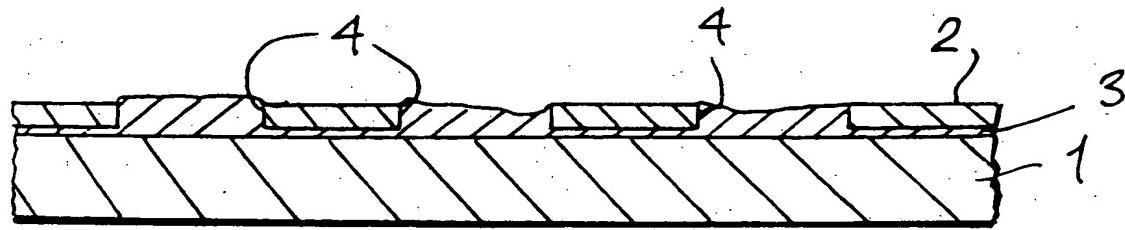


FIG. 1

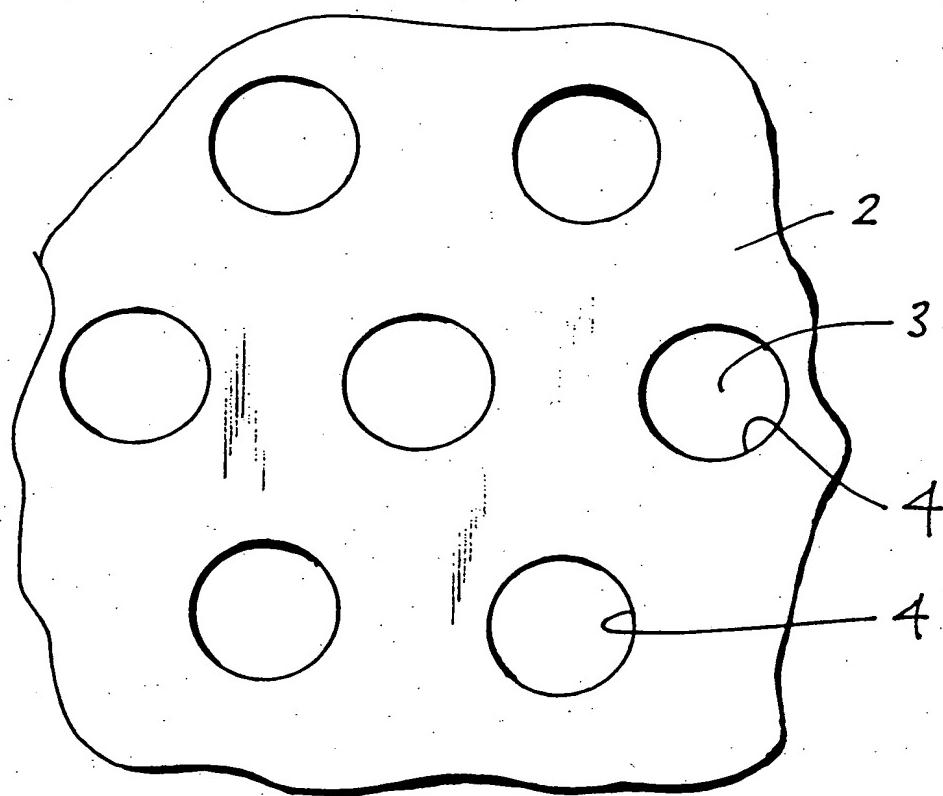


FIG. 2

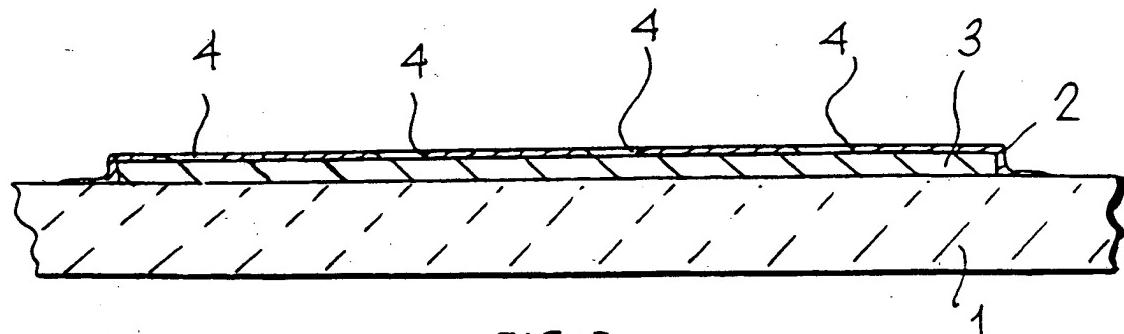


FIG. 3

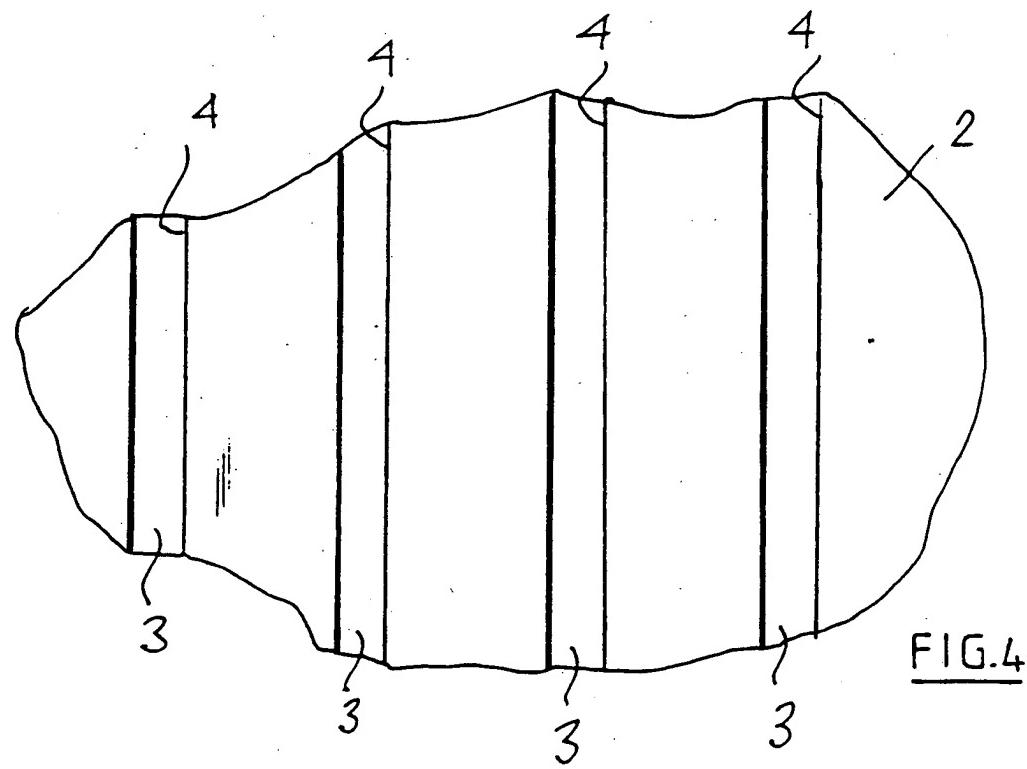


FIG. 4

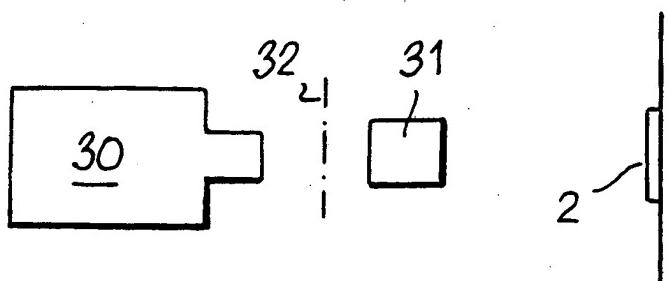
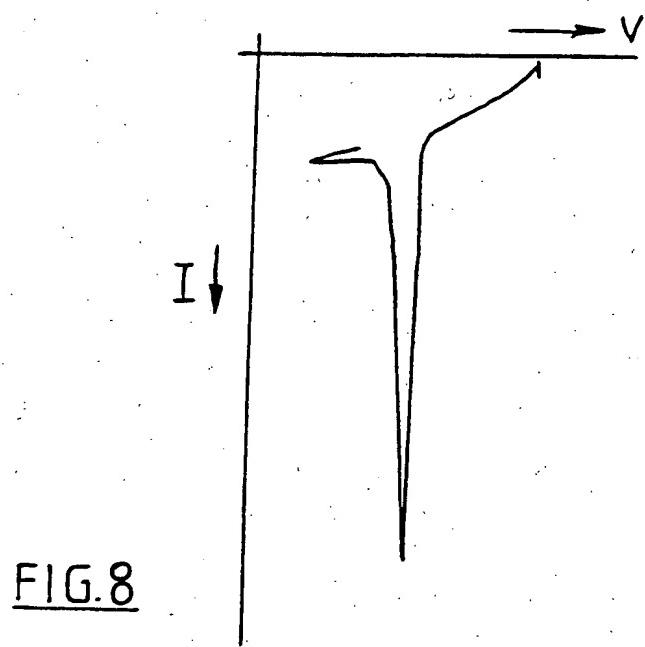
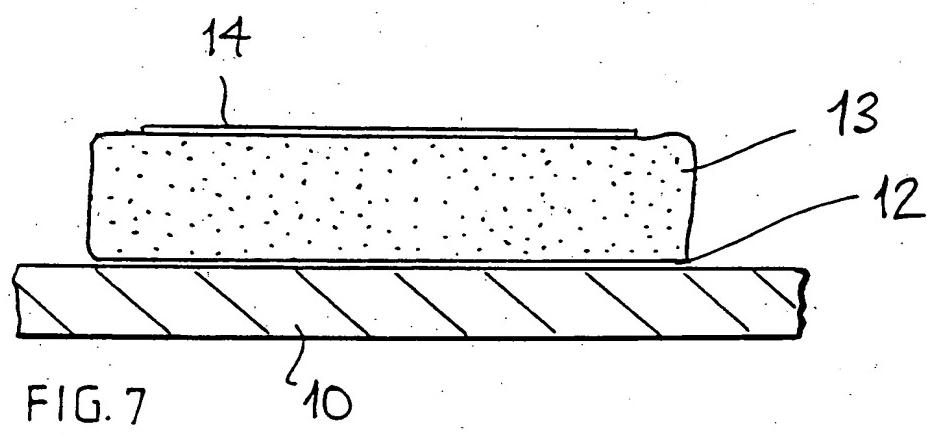
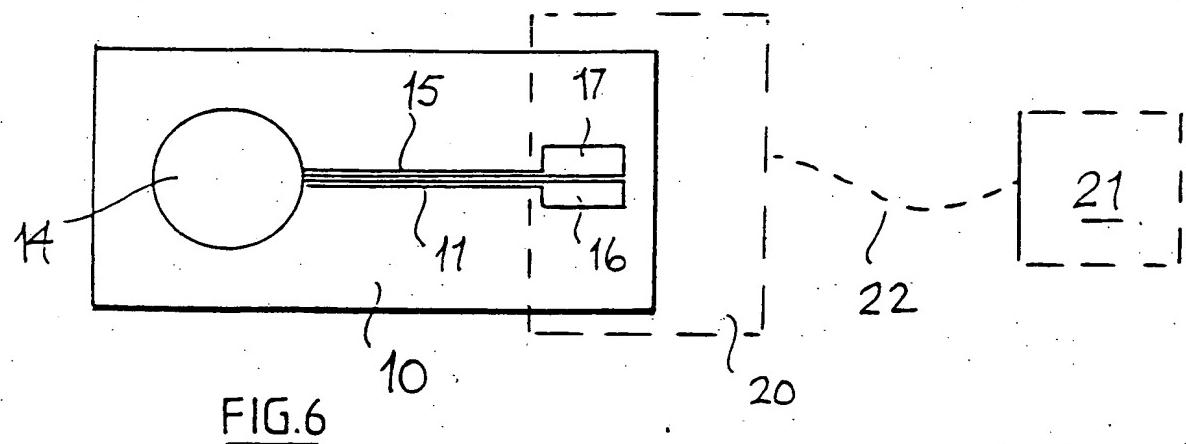


FIG. 5



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